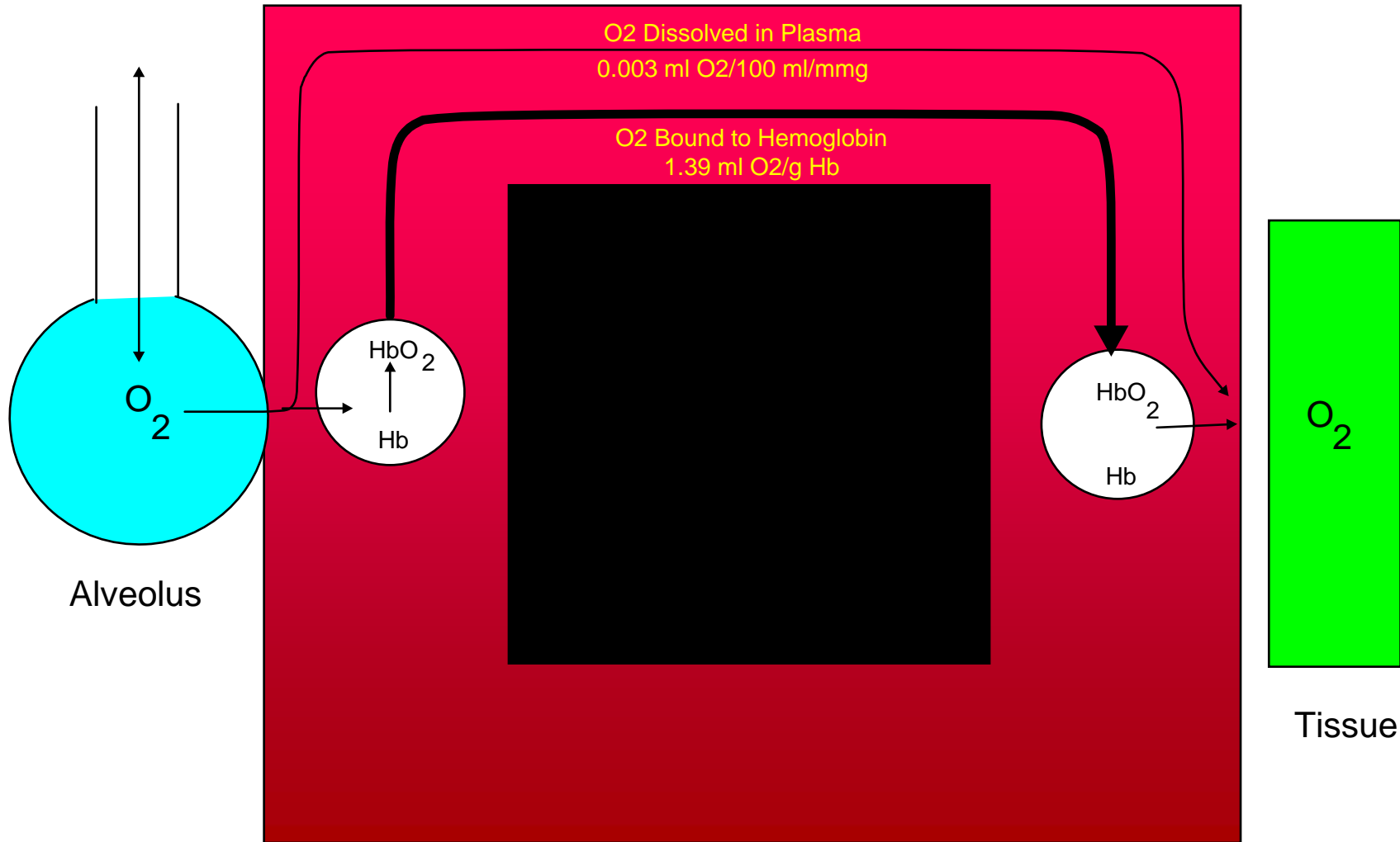


Gas Transport by the Blood

Oxygen Transport



Transport of Dissolved Oxygen

C_{O_2} This is the term for oxygen content.

$$\text{Oxygen solubility} = 0.003 \text{ mlO}_2/\text{dL}/\text{mmHgO}_2 * P_{O_2}$$

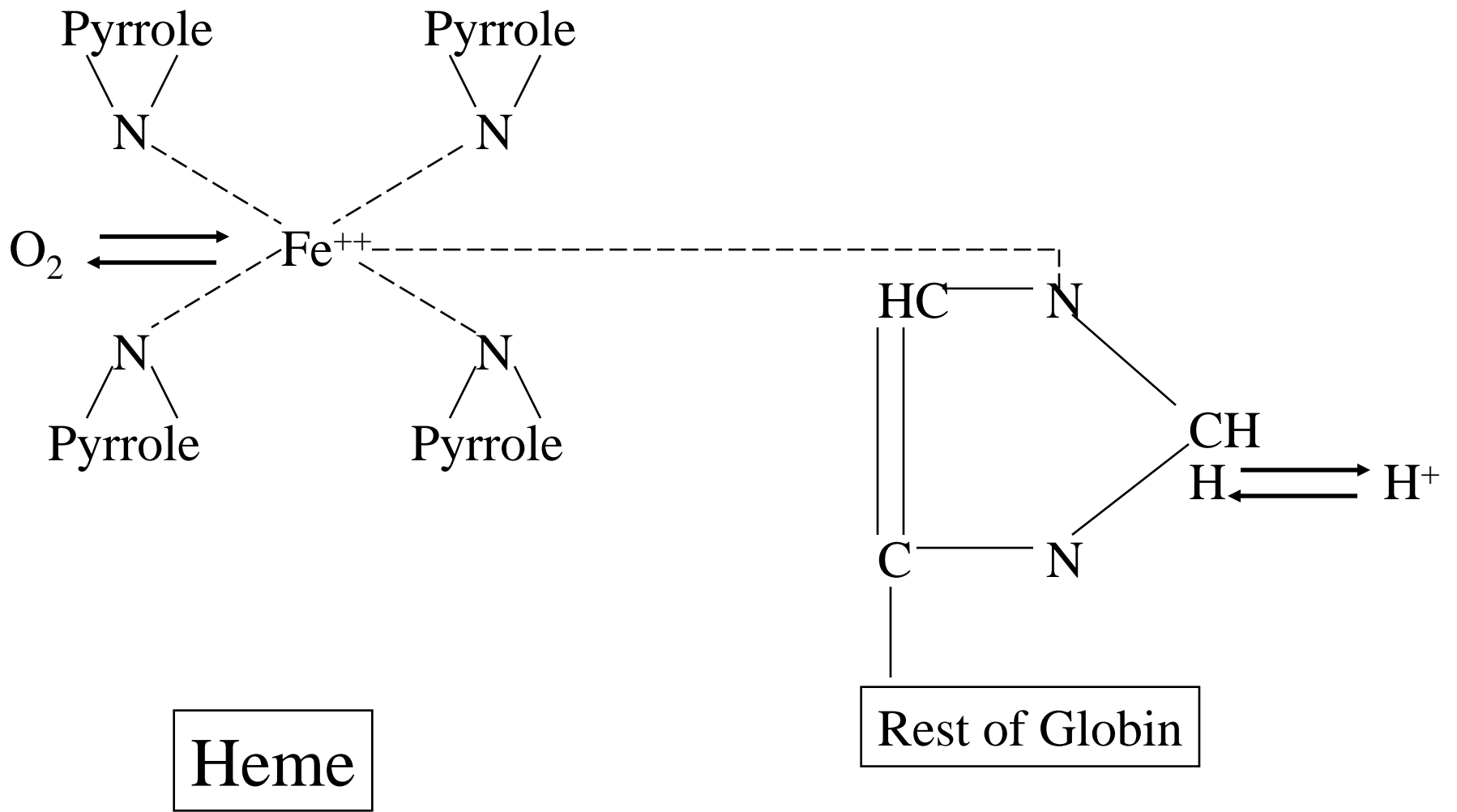
Dissolved oxygen content:

$$P_{aO_2} = 100 \text{ mmHg}$$

$$P_{vO_2} = 40 \text{ mmHg}$$

$$C_{aO_2} = 0.003 \text{ mlO}_2/\text{dL}/\text{mmHgO}_2 * 100 \text{ mmHg}$$

$$C_{aO_2} = 0.3 \text{ mlO}_2/\text{dL blood}$$



Imidazole group on Histidine residue of globin

Hemoglobin Bound Oxygen Content

1 gram HB binds 1.39 mlO₂/dl blood

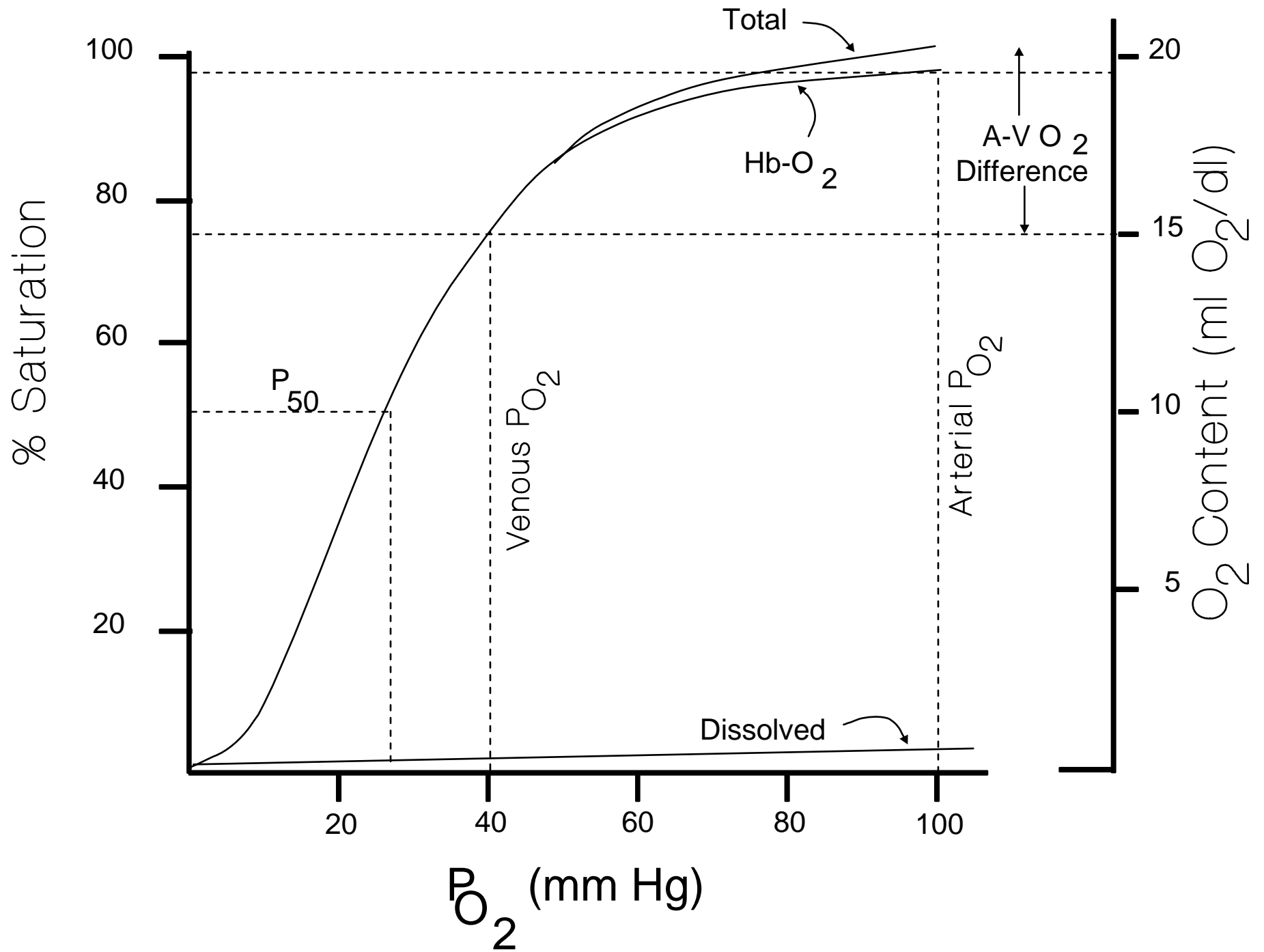
Normal blood [HB] = 15 gm/dL

HB bound oxygen content:

$$C_{O_2} = 1.39 \text{ ml O}_2/\text{gm} * [\text{HB}]$$

$$C_{O_2} = 1.39 \text{ ml O}_2/\text{gm} * 15 \text{ gm/dL}$$

$$C_{O_2} = 20.85 \text{ ml O}_2/\text{dL blood}$$



Total Blood Oxygen Content

$$\text{Normal blood [HB]} = 15 \text{ gm/dL}$$

$$P_{aO_2} = 100 \text{ mmHg}$$

$$P_{vO_2} = 40 \text{ mmHg}$$

$$\text{Oxygen solubility} = 0.003 \text{ mlO}_2/\text{dL}/\text{mmHgO}_2 * P_{O_2}$$

Total blood oxygen content is the sum of HB bound and dissolved:

$$C_{O_2} = (1.39 \text{ ml O}_2/\text{gm} * [\text{HB}] * \% \text{ Sat}) + (0.003 \text{ mlO}_2/\text{dL}/\text{mmHgO}_2 * P_{O_2})$$

$$C_{aO_2} = (1.39 \text{ ml O}_2/\text{gm} * 15 \text{ gm/dL} * 0.97) + (0.003 \text{ mlO}_2/\text{dL}/\text{mmHgO}_2 * 100 \text{ mmHg})$$

$$C_{aO_2} = 20.22 \text{ ml O}_2/\text{dL blood} + 0.3 \text{ ml O}_2/\text{dL blood}$$

$$C_{aO_2} = 20.52 \text{ ml O}_2/\text{dL blood}$$

